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EASTERN UNIVERSITY, SRI LANKA

SECOND EXAMINATION IN SCIENCE - 2000/2001

(May 2001)

PH201 - Atomic Physics and Quantum Mechanics

Time: 02 hours.

Answer All questions.

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2001 / May

1. State the postulates of Bohr theory. Derive an expression for the total energy of the  $n^{\text{th}}$  Bohr's orbit of the Bohr atom and explain its significance. Hence, show that the wavelength of the electromagnetic radiation emitted in a transition between two states of a Bohr atom as

$$\frac{1}{\lambda} = R \left( \frac{1}{n_f} - \frac{1}{n_i} \right),$$

where  $\lambda$  is the wavelength of the radiation,  $R$  is the Redberg constant and  $n_i$  and  $n_f$  are integers.

Deduce the wavelength  $\lambda_\alpha$  of the  $H_\alpha$ -line in the Balmer series of  $H$ -atom as

$$\frac{1}{\lambda_\alpha} = \frac{5R_H}{36},$$

where  $R_H$  is the Redberg constant for  $H$ -atom.

If the wavelength of the  $H_\alpha$ -line in the Balmer series of  $H$ -atom is  $6563\text{\AA}$ , find

- a value for the Redberg constant and
  - the shortest wavelength in the Balmer series limit.
2. Explain briefly the nature of the Zeeman effect in a magnetic field. A hydrogen atom makes the transition from  $n = 2$  to  $n = 1$  state, in which light of frequency  $\nu_0$  is emitted. Show that in a magnetic field  $B$ , the emitted radiation can now have frequencies

$$\nu_0 + \frac{eB}{4\pi m}, \quad \nu_0 \quad \text{and} \quad \nu_0 - \frac{eB}{4\pi m}$$

In the above case calculate the wavelength separation between the two component lines which are observed in Zeeman effect in the magnetic field of  $0.4 \text{ Tesla}$ . The wavelength of the radiation is  $6000\text{\AA}$  and  $\frac{e}{m}$  is  $1.76 \times 10^{11} \text{ Coulomb kg}^{-1}$ .

3. Explain what is meant by Compton effect?

Show that the change in wavelength of a photon subject to Compton scattering by an electron is given by

$$\Delta\lambda = \frac{h}{m_0c} (1 - \cos\phi)$$

where  $\phi$  is the scattering angle and the other symbols have their usual meanings.

In a Compton effect, show that the kinetic energy imparted to the recoiling electron as

$$\frac{hc\Delta\lambda}{\lambda^2}$$

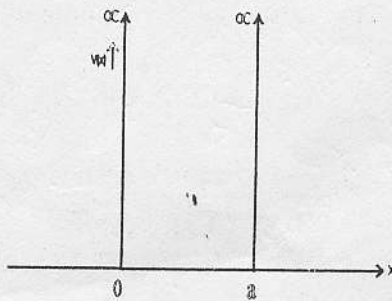
If X-ray of wavelength  $\lambda = 1.0\text{\AA}$  is scattered from a carbon block find

(a) the maximum Compton shift

(b) the kinetic energy imparted to the recoiling electron when the photon recoils at  $90^\circ$  to the incident beam.

4. Write down the time independent Schrödinger equation in a rectangular cartesian coordinate system, for a particle of mass  $m$  and the energy  $E$  moving in a potential  $V$ .

A particle of mass  $m$  and the energy  $E$  moves inside a potential well  $V(x)$  as shown in the figure



$$V(x) = 0 \quad \text{for} \quad 0 \leq x \leq a,$$

$$V(x) = \infty \quad \text{for} \quad x < 0, \text{ and } x > a.$$



- (a) Write down the time independent Schrodinger equation for the motion of the particle.
- (b) State clearly the boundary conditions and the normalization condition for the wavefunction.
- (c) Using the above conditions, show that the wavefunction of the particle as

$$\psi = \sqrt{\frac{2}{a}} \sin\left(\frac{n\pi}{a}\right) x.$$